## MARK SCHEME for the May/June 2015 series

## **0653 COMBINED SCIENCE**

0653/21

Paper 2 (Core Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

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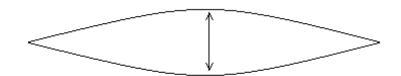
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Page 2		Mark Scheme	Syllabus	Paper
		Cambridge IGCSE – May/June 2015	0653	21
1	(a) (i)	1 proton ; 1 electron ;		[2]
	(ii)	covalent;		[1]
	(iii)	hydrogen + oxygen ; water ;		[2]
	(iv)	heat energy given out / increase in temperature ;		[1]
	(v)	named metal above hydrogen in reactivity series up to and includir above hydrogen in reactivity series ;	ng calcium ;	[2]
	<b>(b)</b> no	ble gas so unreactive (with oxygen) / not flammable ;		[1]
	( <b>c</b> ) C <sub>3</sub>	H <sub>8</sub> ;		[1]
				[Total: 10]
2	(a) (i)	carbon, hydrogen, oxygen ;		[1]
	(ii)	carbon, hydrogen, oxygen ;		[1]
	(b) (i)	X cell membrane ; Y cytoplasm ;		[2]
	(ii)	from alveoli into blood / capillaries ; in blood ; in red cells ; carried by haemoglobin ; any valid reference to diffusion ;		[max 2]
	<b>(c)</b> (e	nergy needed) for contraction of muscles / movement ;		[1]
	(d) (i)	2760 and 2260 ;		[1]
	(ii)	Sarbjit because she used more energy ; she broke down a greater amount of food stores ; (allow ecf if calculation in (i) indicates the wrong girl)		[2]
	(iii)	activities done at different rates owtte ;		[1]
				[Total: 11]

Pa	age 3	3	Mark Scheme Cambridge IGCSE – May/June 2015	Syllabus 0653	Paper 21
3	(a)	wei	ight (accept gravity) ;		[1]
	(b)		o <b>B</b> : accelerating/going faster ; o <b>C</b> : constant speed ;		[2]
	(c)	(OF	stance =) speed x time (in any form) ; R use of area under graph between <b>B</b> and <b>C</b> ) 5 x 30 = 750 (m) ;		[2]
	(d)	(i)	reduces friction(al force) (opposing effect of gravity);		[1]
		(ii)	reduces air resistance (opposing effect of gravity);		[1]
	(e)				[2]
		reg	acceptable equivalent with at least 12 spheres in total gular pattern ; st touching ;		[Total: 9]
4	(a)	(i)	green to yellow / orange / red ;		[1]
	. ,	(ii)	gas dissolves in / reacts with water etc. in atmosphere ; acid rain falls on soil ;		[2]
		(iii)	calcium chloride ; water ;		[2]
	(b)	(i)	decreasing size of pieces decreases time taken/ increases rate/v.v.;		[1]
		(ii)	increasing concentration (of acid) decreases time/ increases rate/v.v. <b>OR</b>		[1]
			increasing temperature decreases time/increases rate/v.v.;		[Total: 7]

Ρ	age 4	4	Mark Scheme	Syllabus	Paper
			Cambridge IGCSE – May/June 2015	0653	21
5	(a)	(i)	arrow correctly drawn from anther of flower <b>A</b> ; to stigma of flower <b>B</b> ; (allow 1 mark if the arrow points to the correct structures but is the	wrong way	<b>[2]</b> round)
		(ii)	large petals ; anthers inside flower ; stigma inside flower ;		[max 2]
	(b)	(i)	no germination at 4°C / in dish <b>3</b> ; no germination when water is absent / in dish <b>2</b> ;		[2]
		(ii)	(light is not needed) no mark because germination took place in dish <b>4</b> ;		[1]
		(iii)	oxygen ;		[1]
					[Total: 8]
6	(a)	(i)	cello ;		[1]
		(ii)	harp ;		[1]
		(iii)	harp ;		[1]
	(b)	(i)			



	or similar diagram to illustrate a vibrating string ;	[1]
(ii)	greater amplitude / owtte ;	[1]
•	ne delay=) distance / speed of sound ; 6 / 330 = 0.2 (s) ;	[2]

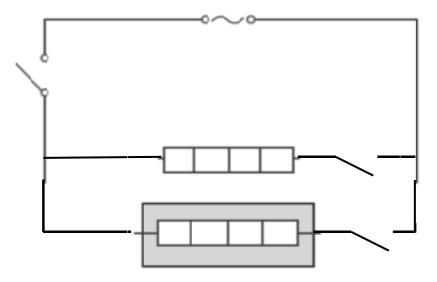
[Total: 7]

(c)

Page 5		Mark Scheme	Syllabus	Paper
		Cambridge IGCSE – May/June 2015	0653	21
7	<b>(a)</b> liqu sol 1 fo			[2]
	(b) (i)	anode cathode ; electrolyte ;		[2]
	(ii)	<b>X</b> on or near left-hand electrode under or just above electrolyte surface ;		[1]
	(iii)	brown / orange / yellow, colouration of, electrolyte/gas ;		[1]
	(c) (i)	(sodium) chloride ;		[1]
	(ii)	(sodium) iodide ;		[1]
	(iii)	trend in reactivity with other halides: $Cl > Br > I / chlorine$ is more reactive than iodine ;		[1]
				[Total: 9]
8	(a) (i)	water ; sugar/glucose ;		[2]
	(ii)	zebra/lion ; lion ;		[2]
		rect arrow drawn from zebra to hyena ; rect arrow drawn from hyena to lion ;		[2]
	(c) (i)	by eating ;		[1]
	(ii)	carbon lost in waste materials / urine / faeces ; carbon lost during respiration as carbon dioxide ; not all the zebra eaten ; not all the zebra digested / absorbed ;		[max 2]
				[Total: 9]

Page 6	Mark Scheme	Syllabus	Paper
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9 (a) (i)	convection ;		[1]
(ii)	conduction ;		[1]
(iii)	any reasonable description of thermal insulation/lagging ; explanation <b>either</b> in terms of reducing thermal energy transfer by <u>conduction</u> through tank wall <b>or</b> the lagging preventing thermal energy transfer by <u>convection</u> ;		[2]

 (b) switches in both heater branches (can be either side of heater); rest of circuit completed properly; (accept any circuit that fulfils the criteria (with or without single switch))



- (c) resistance of water heater less than that of warm air heater ;
  p.d. same across both, so current twice / higher,
  and so resistance must be half / lower ;
  (or vice versa)
- (d) damaged insulation ; accept water leak / dampness the heater is not earthed ;

[max 1]

[3]

[2]

[Total: 10]